

0957-4166(95)00402-5

Combined Microbial Oxidation and Reduction: a New Approach to the High-yield Synthesis of Homochiral Unsaturated Secondary Alcohols from Racemates

Giancarlo Fantin, Marco Fogagnolo, Pier Paolo Giovannini, Alessandro Medici,*
and Paola Pedrini

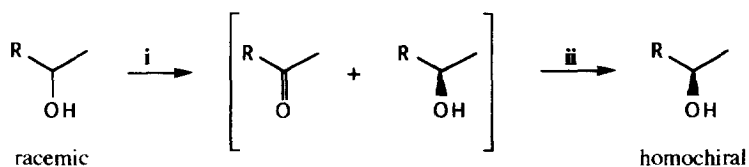
Dipartimento di Chimica, Università di Ferrara, Via L. Borsari 46, I-44100 Ferrara, Italy

Key Words: Microbial oxidation, Microbial reduction, *Bacillus stearothermophilus*, *Yarrowia lipolytica*, Unsaturated secondary R-alcohol

Abstract: The oxidation of racemic secondary alcohols with *Bacillus stearothermophilus* followed by reduction of the mixture with *Yarrowia lipolytica* to afford high yields of the enantiomerically pure R-alcohols **1a,b,d** is described. Comparisons with *Yarrowia lipolytica* reduction, *Bacillus stearothermophilus* oxidation, and the contemporary use of the two microorganisms have been made.

Enantiomerically pure secondary alcohols are very common as pheromones,¹ aroma, flavour enhancing compounds, intermediates and chiral auxiliaries in organic synthesis.² The methodologies to obtain these chiral compounds are the microbial or enzymatic reduction of the corresponding ketones and the kinetic resolution of the racemic alcohols *via* esterification or hydrolysis of the corresponding esters with lipases. A broad range of ketones are reduced with Baker's yeast³ or using purified alcohol dehydrogenases.⁴ During the course of the reaction with Baker's yeast the S-alcohol was almost exclusively obtained, while the enzyme delivers the hydride either from the *si*-side (i.e. *Lactobacillus kefir* alcohol dehydrogenase⁵) or *re*-side (i.e. *Thermoanaerobium brockii* alcohol dehydrogenase⁶) of the ketone to give (R)- or (S)-alcohols, respectively. For most cases, the stereochemical course of the reaction, which is mainly dependent on the steric requirements of the substrate, may be predicted from a simple model which is generally referred to as "Prelog's rule."⁷ The majority of the commercially available dehydrogenases used for the stereospecific reduction of ketones and the majority of microorganisms follow the Prelog's rule giving the S-alcohols. However, the enzymatic resolution of the racemic alcohols with lipases,⁸ when a high enantioselectivity is exhibited, stops at 50% conversion stage.⁹ In this field our previous work described an alternative method for the kinetic resolution of secondary alcohols *via* oxidation with Baker's yeast¹⁰ and *Bacillus stearothermophilus*^{11,12} affording R-enantiomer in high enantiomeric excess and only recently an "anti-Prelog" behaviour (formation of the R-alcohol) is reported for the microbial reduction of various carbonyl compounds with *Yarrowia lipolytica* strains, a recent yeast species.¹³ In this paper we describe a new approach to the high-yield synthesis of homochiral secondary R-alcohols starting from the corresponding racemates which consists in the following pathway: i) kinetic resolution *via*

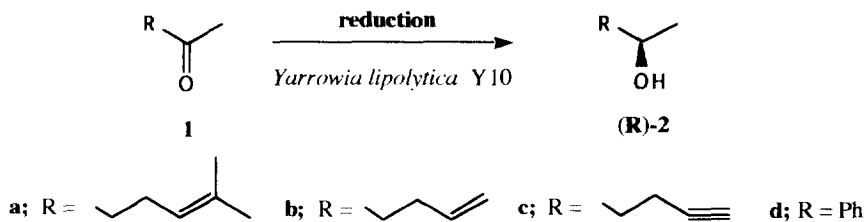
oxidation with *Bacillus stearothermophilus*; ii) reduction of the obtained oxidation mixture with *Yarrowia lipolytica* (Scheme 1).



Scheme 1

This methodology is compared with the other classical approaches: i) reduction with *Yarrowia lipolytica* of the corresponding ketones; ii) kinetic resolution via oxidation with *Bacillus stearothermophilus* of the racemic alcohols. On the whole the combination of the oxidation followed by the reduction is a choice when the reduction is not so efficient and considering that the kinetic resolution stops at 50% conversion.

The results of the reduction of the ketones **1a-d** (Scheme 2) with *Yarrowia lipolytica* Y10¹⁴ are summarized in Table 1.



Scheme 2

Table 1. Reduction of the ketones **1a-d** with *Yarrowia lipolytica* Y10.

ketone	culture media (time h)	alcohol (yield %)	ee % ^a (abs. conf.) ^b
1a	synthetic (24)	2a (75)	95 (R)
1b	synthetic (24)	2b (26)	68 (R)
1b	saboraud (24)	2b (30)	48 (R)
1c	synthetic (24)	2c (83)	94 (S)
1d	synthetic (24)	2d (26)	62 (R)
1d	saboraud (24)	2d (25)	100 (R)

^a Determined by GLC on chiral column containing *n*-pentyl-dimethyl- β -cyclodextrin. ^b Determined comparing the sign of the optical rotation with that of the literature (ref. 15).

The substrate was added to a growing culture of *Yarrowia lipolytica* as a concentrated solution in DMSO and the incubation was prolonged for 24 h at 28°C. 5-Methyl-5-hepten-2-one **1a** and 5-hexyn-2-one **1c** are reduced

with good yields and enantiomeric excesses to R-sulcatol **2a** (75%, ee 95%) and S-hexynol **2c** (83%, ee 94%), respectively. The anomalous behaviour of **1c** in respect to a series of nonconjugated unsaturated ketones is also observed in the reduction of **1c** with *Thermoanaerobium brockii* alcohol dehydrogenase where the ketone gives the R- instead of the S-enantiomer.¹⁵ The reductions of the 5-hexen-2-one **1b** and of the acetophenone **1d** produce the corresponding R-alcohols **2b** and **2d** with quite good ee (48-100%) but with low yields (25-30%) also varying the culture media.

The R-alcohols are also obtained by the kinetic resolution of the racemic unsaturated secondary alcohols **1a-d** via oxidation with *Bacillus stearothermophilus* (Scheme 3) and the results are summarized in Table 2.

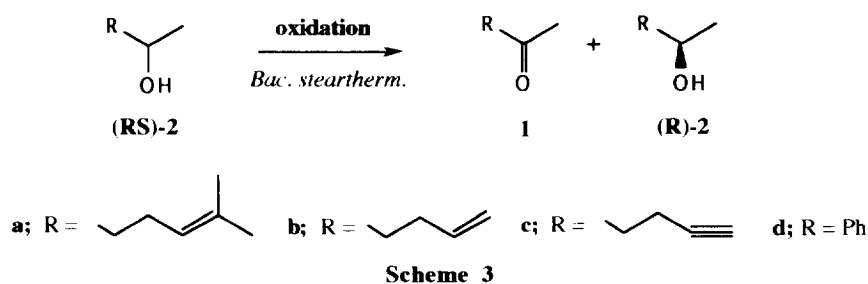


Table 2. Kinetic resolution of unsaturated racemic alcohols **1a-d** with *Bacillus stearothermophilus*.

Alcohol	Products ^a	% ratio ^b	Time (h)	% ee ^b (abs. conf.) ^c
2a	1a	52	3	
	2a	48		98 (R)
2b	1b	51	4	
	2b	49		100 (R)
2c	1c	51	20	
	2c	49		99 (R)
2d	1d	53	6	
	2d	47		85 (R)

^a The mixture of the products **1** and **2** was purified on a short column of silica gel (recovered material 90%). ^b Determined by GLC on chiral column containing *n*-pentyl-dimethyl- β -cyclodextrin. ^c Determined comparing the sign of the specific rotation with that of the literature (ref. 15).

The substrate was added to a growing culture as a concentrated solution in DMSO and the incubation was prolonged for the appropriate time at 39 °C. All the reactions afforded the ketones **1a-d** leaving the unreacted R-enantiomers of the corresponding alcohols **2a-d** (47-49%) with good enantiomeric excesses (85-100%).

On the basis of these results considering that in some cases the microbial reduction of the ketones is not so efficient and the oxidation stops at 50% conversion, and since both the reactions bring to the R-enantiomer, we combined the oxidation and the reduction in order to obtain higher yields of homochiral R-alcohols starting from the corresponding racemates. The methodology consists in carrying out the kinetic resolution of the racemic

alcohol by oxidation. For the growth of *Bac. stearotherm.* a poorer nutrient broth was used. In one case the same nutrient broth has been utilized for the oxidation of the alcohols **2** (method A), in another one (method B) the cells of *Bac. stearotherm.* were centrifuged and suspended in phosphate buffer. The oxidation reactions, in both cases, are carried out at 39°C for 5 h and then the cells were centrifuged and the surmatant, containing the mixture of ketone **1** and the unreacted alcohol **2**, was added to grown up (see experimental: method A) and centrifuged cells of *Yarrowia lipolytica*. The reduction was continued for a further 24 h at 28°C (Scheme 4).

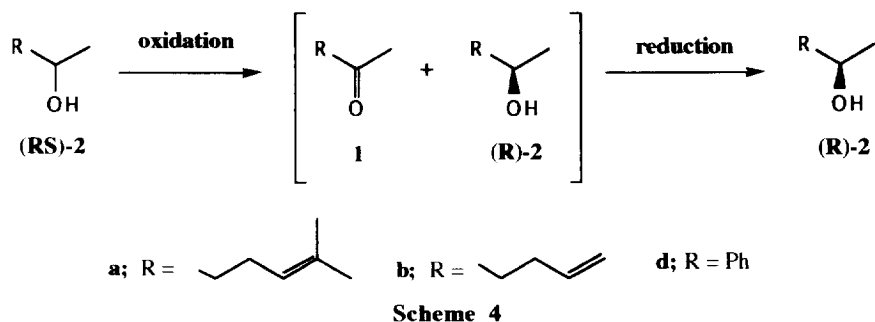


Table 3. Combined oxidation and reduction of the racemic alcohols **1a, b, d**.

(R,S)-2	Method ^a	Oxidation ^b -Reduction ^c		
		1 (yield %)	2 (yield %)	ee% ^d (abs. conf.)
2a	A	1a (13)	2a (87)	100 (R)
2a	B	1a (9)	2a (91)	100 (R)
2a	C ^e	1a (18)	2a (82)	90 (R)
2b	A	1b (0)	2b (100)	100 (R)
2b	B	1b (9)	2b (91)	100 (R)
2b	C ^e	1b (15)	2b (85)	40 (R)
2d	A	1d (12)	2d (88)	40 (R)
2d	B	1d (20)	2d (80)	85 (R)
2d	C ^e	1d (0)	2d (100)	0

^a The reaction is carried out: in the nutrient broth of *Bacillus stearothermophilus* (method A); in phosphate buffer pH 7.2 (method B); in the mixing of the nutrient broth of *Bacillus stearothermophilus* and *Yarrowia lipolytica* (method C). ^b The crude reaction mixture of the oxidation with *Bacillus stearothermophilus* is centrifuged and the surmatant is added to the grown cells of *Yarrowia lipolytica*. ^c The *Yarrowia lipolytica* cells, grown up in a synthetic medium (see experimental), are centrifuged and the reduction is carried out with the surmatant obtained from the oxidation. ^d Enantiomeric excesses are calculated from glc analysis on chiral column containing *n*-pentyl-dimethyl- β -cyclodextrin. ^e The reaction is carried out with mixed grown up cells of *Bacillus stearothermophilus* and *Yarrowia lipolytica*.

The results of the combined oxidation and the reduction are reported in Table 3.

The oxidation of 6-methyl-5-hepten-2-ol (sulcatol) **2a** followed by reduction afforded in very high yield (87-91%) the pure R-enantiomer (method A and B). Moreover, pure (R)-5-hexen-2-ol **2b** is obtained in quantitative

yield (method A). It is worth mentioning that in the case of the R-sulcatol **2a** the combination of the two reactions affords only a small improvement, while for the hexenol **2b** this methodology is more efficient than the single reactions.

High yield (80%) and good enantiomeric excess (85%) are also obtained with 1-phenylethanol **2d** carrying out the reactions in phosphate buffer while the method A (nutrient broth of *Bac. stearotherm.*) affords poorer enantiomeric excess (40%). Obviously this approach is not used when the reduction affords the S-enantiomer (i.e. 5-hexyn-2-one **1c**).

As shown in Table 3 (redox), a parallel experiment was made mixing *Bacillus stearothermophilus* and *Yarrowia lipolytica*, grown up in the appropriate culture media, and adding the alcohols **2a, b, d**. The incubation was continued for 24 h at 28°C. Worst results are obtained for the racemic alcohols **2a** (85% yield, ee 90%) and **2b** (85% yield, ee 40%), while the alcohol **2d** is completely not resolved.

In conclusion, matching oxidation with *Bacillus stearothermophilus* and reduction with *Yarrowia lipolytica*, it is possible to obtain homochiral R-alcohols starting from the racemate in high yields. This methodology is a third pathway and is a choice when the reduction with *Yarrowia lipolytica* is not so efficient and/or because the oxidation stops at 50% conversion.

Experimental

Optical rotations were measured on a Perkin Elmer Model 241 polarimeter. Gas chromatographic analyses were performed on a Carlo Erba GC 6000 Vega series 2. The ketones **1a** (Fluka) and **1d** (Fluka) and the alcohols **2a** (Aldrich) and **2d** (Fluka) are commercially available. The alcohols **2b**¹⁶ and **2c** were obtained by reduction with sodium borohydride of the corresponding ketones **1b**¹⁷ and **1c**.¹⁸

5-Hexyn-2-ol **2c** showed the following: oil; ¹H NMR (200 MHz) δ 1.18 (d, 3 H, $J = 5.9$ Hz), 1.62 (q, 2 H, $J = 6.7$ Hz), 1.94 (t, 1 H, $J = 2.5$ Hz), 2.17 (br s, 1 H), 2.28 (dt, 2 H, $J = 2.5$ and 6.7 Hz), 3.91 (m, 1 H).

Separation by GLC. Enantiomer separation was obtained on a Megadex 5 column (25 m X 0.25 mm) containing *n*-pentyl dimethyl β -cyclodextrin in OV 1701 from Mega s.n.c. For compound **1a**: helium 0.8 atm; temp. 80-200°C (1.5°C/min); retention time in min (after acetylation with acetic anhydride and pyridine) 3.88 **1a**, 5.96 (S)-**2a**, 6.72 (R)-**2a**. For compound **1b**: helium 0.9 atm; temp. 80°C; retention time in min 4.76 **1b**, 8.7 (S)-**2b**, 8.95 (R)-**2b**. For compound **1c**: helium 0.9 atm; temp 100-200°C (2°C/min); retention time in min 3.15 **1c**, 4.41 (S)-**2c**, 4.58 (R)-**2c**. For compound **1d**: helium 0.8 atm; temp. 100-200°C (2°C/min); retention time in min 10.21 **1d**, 14.84 (R)-**2d** 15.43 (S)-**2d**. The absolute configurations of the compounds were determined to be R comparing the sign of their specific rotation with those of the literature: (R)-6-methyl-5-hepten-2-ol **2a**¹⁵ $[\alpha]_D = -14.5$ (c 1.3, EtOH); (R)-5-hexen-2-ol **2b**¹⁵ $[\alpha]_D = -12.1$ (c 4.6, CHCl₃); (R)-5-hexyn-2-ol **2c**¹⁵ $[\alpha]_D = -25.9$ (c 3.5, CHCl₃); (R)-1-phenylethanol **2d**¹¹ $[\alpha]_D = 41$ (c 5.1, CHCl₃).

Microbial Reduction with *Yarrowia lipolytica* Y10 of the Ketones 1a-d. General Procedure.

A synthetic culture medium is prepared adding to 1 L of water glucose (50 g), (NH₄)₂SO₄ (5g), KH₂PO₄ (2 g), CaCl₂ (0.25 g), MgSO₄·7H₂O (0.25 g), inositol (25 mg), H₃BO₃ (1 mg), ZnSO₄ (1 mg), MnCl₂ (1 mg), FeCl₂ (0.5 mg), CuSO₄ (0.1 mg), tiamine (0.3 mg), biotine (0.025 mg), calcium pantothenate (0.3 mg), pyridoxine (0.3 mg) and nicotinic acid (0.3 mg). The culture of *Yarrowia lipolytica* Y10 was set up in 500 mL conical flask containing the synthetic medium (100 mL) and stirred at 28°C for 48 h.

To the resulting suspension of grown cells the ketone **1** (0.2 g) in DMSO (1 mL) was added. After 24 h the reaction mixture was extracted with diethyl ether (100 mL) with a continuous liquid-liquid extractor, dried over anhydrous Na₂SO₄ and concentrated under vacuum. The alcohols were purified by column chromatography (silica gel, petroleum ether/diethyl ether 80:20). The yields are reported in Table 1 and the enantiomeric excesses are calculated by GLC analysis on chiral column.

Microbial oxidations with *Bacillus stearothermophilus* of the Alcohols 2a-d. General procedure. The nutrient broth was prepared dissolving bactotryptone (2 g), yeast extract (1 g) and sodium oxalate (0.3 g) in distilled water (100 mL). The culture of *Bacillus stearothermophilus* ATCC2027 was set up in 500 mL conical flask containing the nutrient broth (100 mL) and stirred at 39°C for 48 h. To the resulting suspension of grown cells the racemic alcohol **2** (0.3 g) in DMSO (1 mL) was added. After the appropriate time (see Table 2), the reaction mixture was extracted with diethyl ether (100 mL) with a continuous liquid-liquid extractor, dried over anhydrous Na₂SO₄ and concentrated under vacuum. The products were purified by column chromatography (silica gel, petroleum ether/diethyl ether 80:20) to give the ketones and the unreacted alcohols (total yields 90%). The **2:1** molar ratios and the enantiomeric excesses from GLC are given in Table 2.

Combined oxidations and reductions of Alcohol 2a,b,d. General procedure.

Method A. The nutrient broth for the growth of *Bacillus stearothermophilus* was prepared dissolving glucose (1 g), bactotryptone (5 g), yeast extract (1 g) in water (1 L). A sterilized nutrient broth (50 mL) was inoculated with a loopful of the *Bacillus stearothermophilus* and incubated for 48 h at 39°C under stirring. To the resulting suspension of grown cells the racemic alcohol **2** (0.1 g) in DMSO (1 mL) was added. After stirring for a further 5 h at 39°C, the cells were centrifuged. The supernatant was added to the centrifuged cells of *Yarrowia lipolytica* grown for 48 h at 28°C in 50 mL of a stirred synthetic medium (see above for reduction).

The reduction with *Yarrowia lipolytica* was continued for 24 h at 28°C under stirring. The reaction mixture was centrifuged and the supernatant was extracted with diethyl ether (50 mL) with a continuous liquid-liquid extractor, dried over anhydrous Na₂SO₄ and concentrated under vacuum. The products were purified by column chromatography (silica gel, petroleum ether/diethyl ether 80:20) to obtain the alcohols and the unreacted ketones. The yields and the enantiomeric excesses (from GLC) are listed in Table 3.

Method B. The oxidation and the subsequent reduction were carried as above (method A) but the grown up cells of *Bacillus stearothermophilus*, after centrifugation, were suspended in phosphate buffer at pH 7.2 (50 mL) and the racemic alcohol **2** (0.1 g) in DMSO (1 mL) was added. The results are summarized in Table 3.

Method C. The suspension of the grown up cells (method A) of *Bacillus stearothermophilus* (50 mL) and the suspension of grown up cells (in synthetic medium) of *Yarrowia lipolytica* (50 mL) were mixed and the racemic alcohol **2** (0.1 g) was added. After 24 h stirring at 27°C the reaction mixture was centrifuged and the supernatant was extracted with diethyl ether (50 mL) with a continuous liquid-liquid extractor, dried over anhydrous Na₂SO₄ and concentrated under vacuum. The products were purified by column chromatography (silica gel, petroleum ether/diethyl ether 80:20) to obtain the alcohols and the unreacted ketones (Table 3).

References

1. Mori, K. *Tetrahedron* **1989**, *45*, 3233.
2. Stinson, S. C. *Chem. Eng. News* **1992**, *46*. Morrison, J. D. *Asymmetric Synthesis*; Academic Press: 1984.
3. (a) Servi, S. *Synthesis* **1990**, *1*. (b) Csuk, R.; Glanzer, B. I. *Chem. Rev.* **1991**, *91*, 49.
4. Lemiere, G. L. *Enzymes as Catalysts in Organic Synthesis*; Schneider, M. P., Ed.; D. Reidel Publishing: Dordrecht, Holland, 1986.
5. Bradshaw, C. W.; Hummel, W.; Wong, C-H. *J. Org. Chem.* **1992**, *57*, 1532.
6. Keinan, E.; Hafeli, E. K.; Seth, K. K.; Lamed, R. *J. Am. Chem. Soc.* **1986**, *108*, 162.
7. Faber, K. *Biotransformations in Organic Chemistry*; Springer-Verlag: 1992.
8. (a) Burgess, K.; Jennings, L. D. *J. Am. Chem. Soc.* **1991**, *113*, 6129. (b) Morgan, B.; Oehlschlager, A. C.; Stokes, T. M. *J. Org. Chem.* **1992**, *57*, 3231.
9. Azerad, R. *Bull. Soc. Chim. Fr.* **1995**, *132*, 17.
10. Fantin, G.; Fogagnolo, M.; Medici, A.; Pedrini, P.; Poli, S.; Sinigaglia, M. *Tetrahedron Lett.* **1993**, *34*, 883.
11. Fantin, G.; Fogagnolo, M.; Medici, A.; Pedrini, P.; Poli, S.; Gardini, F. *Tetrahedron Asymm* **1993**, *4*, 1607.
12. Fantin, G.; Fogagnolo, M.; Medici, A.; Pedrini, P.; Rosini, G. *Tetrahedron Asymm.* **1994**, *5*, 1635. Fantin, G.; Fogagnolo, M.; Giovannini, P. P.; Medici, A.; Pedrini, P.; Poli, S. *Tetrahedron Lett.* **1995**, *36*, 441.
13. Fantin, G.; Fogagnolo, M.; Giovannini, P. P.; Medici, A.; Pedrini, P.; Gardini, F.; Lanciotti, R. *Tetrahedron Lett.*, submitted.
14. *Yarrowia lipolytica* Y10 belongs to DPVA (Dipartimento di Protezione e Valorizzazione Agroalimentare, University of Bologna, Italy) collection.
15. Keinan, E.; Seth, K. K.; Lamed, R.; Ghirlando, R.; Singh, S. P. *Biocatalysis* **1990**, *3*, 57.
16. Friederang, A. W.; Tarbell, D. S. *J. Org. Chem.* **1968**, *33*, 3797. Ragonnet, B.; Santelli, M.; Bertrand, M. *Helv. Chim. Acta* **1974**, *57*, 557.
17. Shono, T.; Nishiguchi, I.; Ohmizu, H.; Mitani, M. *J. Am. Chem. Soc.* **1978**, *100*, 545.
18. Crombie, L.; Hemesley, P.; Pattenden, G. *J. Chem. Soc.(C)* **1969**, 1016.

(Received in UK 26 September 1995)